

Synthesis of 2-Substituted-2,3-dihydro-5-benzoyl-1*H*-1,3,2-benzodiazaphosphole 2-Oxides/Sulfides

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ABSTRACT: Syntheses of 2-aryloxy/2-chloro ethoxy-2,3-dihydro-5-benzoyl-1*H*-1,3,2-benzodiaza-phosphole 2-oxides **3a–h** were accomplished by reactions of equimolar quantities of 3,4-diaminobenzophenone (**1**) with various aryl/chloroethoxy phosphorodichloridates **2a–g** and **2h** in the presence of triethylamine at 50–60°C. Compounds **3i–k** were prepared by reacting 3,4-diaminobenzophenone (**1**) with aryl thiophosphorodichloridates **2i–k** under similar conditions. They were characterized by IR, ¹H, ¹³C, and ³¹P NMR spectral data. Some of these products possessed significant antimicrobial activity © 2002 Wiley Periodicals, Inc. *Heteroatom Chem* 13:340–345, 2002; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.10044

INTRODUCTION

Intensive investigations of compounds containing P–N linkages by numerous research groups have led to many interesting and far reaching developments [1]. Organophosphorus compounds containing a benzodiazaphosphole ring have acquired much importance because of their insecticidal, bactericidal, antiviral, and anticarcinogenic properties [2–4]. The search for a satisfactory drug for cancer also produced many new phosphorus heterocyclic

compounds [5,6], and many of them possessed significant antitumor activity [6,7]. In continuation of our interest in the syntheses of new phosphorus heterocycles, preparation of the title compounds was accomplished.

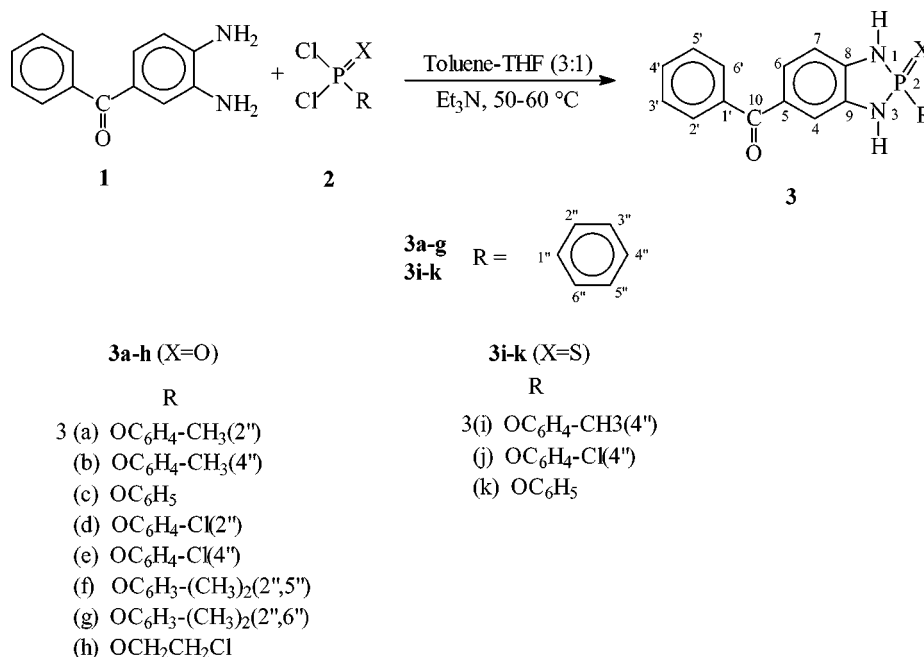
RESULTS AND DISCUSSION

Syntheses of 2-aryloxy-/2-chloroethoxy-2,3-dihydro-5-benzoyl-1*H*-1,3,2-benzodiaza-phosphole 2-oxides **3a–h** and syntheses of 2-substituted 2,3-dihydro-5-benzoyl-1*H*-1,3,2-benzodiazaphosphole 2-sulfides **3i–k** were achieved by the cyclocondensation of equimolar quantities of 3,4-diaminobenzophenone (**1**) with aryl/chloroethoxy phosphorodichloridates **2a–h** and aryl thiophosphorodichloridates **2i–k** in dry toluene–tetrahydrofuran mixture (3:1) in the presence of triethylamine at 50–60°C (Scheme 1). Purifications of **3a–h** and **3i–k** were achieved by filtering off of the triethylamine hydrochloride, evaporation of the filtrate, washing of the residue with water and recrystallization of the solid products from aqueous 2-propanol. Their structures were established by elemental analyses, IR, ¹H, ¹³C, and ³¹P NMR spectral data (Tables 1–3).

The IR spectra of **3a–h** and **3i–k** (Table 1) showed bands at 3230–3372 (P–NH), 1617–1644 (C=O), 1221–1290 (P=O), 768–797 (P=S), 1162–1221, and 952–975 (P–O–C_{aro}), 1080 and 907 (P–O–C_{ali}) [8–11].

In their ¹H spectra (Table 2), a singlet for H-4 was observed at δ 6.73–7.32 for **3**. The doublets at

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SCHEME 1

δ 6.26–7.08 ($J = 8.1$ – 8.5 Hz) are for H-6. The other doublets found at δ 6.89–7.68 ($J = 8.1$ – 8.5 Hz) are attributed to H-7. The aryloxy moiety protons of **3a–h** and **3i–k** exhibited multiplets in the range of δ 6.98–7.84. The chloroethoxy group protons in **3h** appeared as a multiplet at δ 3.42–4.22. The signals for C-5 benzoyl protons in **3** were found as a complex multiplet at δ 6.82–7.84. The broad signals at δ 5.24–6.32 are for amidic protons which are confirmed by a D₂O exchange.

The ¹³C NMR chemical shifts (Table 3) were recorded for some members of the title compounds (**3a–d**, **3g**, **3i**, and **3j**). Patterns for C-4 and C-7 were doublets at 112.67–116.12 ($J = 12.0$ – 12.8 Hz) and 116.22–122.20 ($J = 12.0$ – 12.8 Hz). The chemical shifts at 138.98–140.12 and 125.52–127.31 were assigned for C-5 and C-6, respectively. In the case of C-1', C-(2',6'), C-4', and C-(3',5'), the order of chemical shifts was 138.24–139.78, 128.92–129.49, 131.12–131.79, and 128.94–129.48. The nitrogen-bearing C-8 and C-9 exhibited signals downfield at 141.73–144.73 and 138.98–139.78. The signals for C-1' appeared at 149.86–153.25. The C-2' and C-6' methyl groups in **3a** and **3g** appeared upfield at 16.53 and 16.91 ppm, respectively, because of their γ -interaction with exocyclic oxygen [12–14]. The signals for the carbonyl group (C-10) occurred at 193.74–195.04. The ¹³C NMR chemical shifts for **3e**, **3f**, **3h**, and **3k** were not assignable because of the poor quality of the spectrum since the compounds are meagerly soluble in DMSO.

The ³¹P NMR signals appeared in the range of –6.30 to 15.88 ppm from 85% H₃PO₄ [15–17].

ANTIMICROBIAL ACTIVITY

The compounds **3a–e** and **3i** were screened for antifungal activity on *Aspergillus flavus*, *Penicillium notatum*, *Helminthosporium anomalum*, and *Fusarium oxysporum*, according to the literature technique [18], by using Griseofulvin as a standard at two different concentrations 10 and 20 μ g. Most of the compounds exhibited significant toxicity (Table 4) against all fungi. Antibacterial activity was also assessed using the reported procedure [19], on *Bacillus subtilis* and *Klebsiella pneumoniae* (Table 4). Compounds **3d**, **3e**, and **3i** displayed strong action against the above bacteria.

EXPERIMENTAL

All melting points were determined on a Mel-Temp apparatus and are uncorrected. Elemental analyses were performed by the Central Drug Research Institute, Lucknow, India. The purity of the compounds was checked by thin-layer chromatography (TLC; Silica gel 1H, BDH, hexane/ethyl acetate 2:3). All IR spectra were recorded as KBr pellets on a Perkin-Elmer 1483 unit. The ¹H and ¹³C NMR spectra were taken on Varian Gemini 300 and 400 MHz NMR spectrometers operating at 300 and 400 MHz, respectively. The ³¹P NMR spectra were taken on a

TABLE 1 Physical and IR Data of Compounds 3^a

Compound	M.P. (°C)	Yield (%)	Mol. Formula	Analysis Found (Required) (%)			IR (cm ⁻¹)					
				C	H	N	C=O	P=O	P=S	P-NH	P-O-C (aro)	
											P-O	O-C
3a	168–170 (d)	58	C ₂₀ H ₁₇ N ₂ PO ₃	65.78 (65.93)	4.62 (4.70)	7.64 (7.68)	1621	1290	–	3354	958	1183
3b	140–142 (d)	54	C ₂₀ H ₁₇ N ₂ PO ₃	65.82 (65.93)	4.59 (4.70)	7.65 (7.68)	1617	1279	–	3358	960	1192
3c	152–154 (d)	52	C ₁₉ H ₁₅ N ₂ PO ₃	65.05 (65.14)	4.26 (4.31)	8.07 (7.99)	1618	1258	–	3346	954	1196
3d	68–70 (d)	45	C ₁₉ H ₁₄ N ₂ PO ₃ Cl	59.26 (59.31)	3.78 (3.66)	7.36 (7.28)	1628	1261	–	3372	955	1190
3e	76–78 (d)	48	C ₁₉ H ₁₄ N ₂ PO ₃ Cl	59.23 (59.31)	3.72 (3.66)	7.22 (7.28)	1619	1254	–	3366	952	1192
3f	280–282 (d)	60	C ₂₁ H ₁₉ N ₂ PO ₃	66.62 (66.66)	5.12 (5.06)	7.46 (7.40)	1642	1232	–	3280	965	1202
3g	293–295 (d)	57	C ₂₁ H ₁₉ N ₂ PO ₃	66.75 (66.66)	5.18 (5.06)	7.32 (7.40)	1640	1221	–	3286	936	1162
3h	146–148 (d)	52	C ₁₅ H ₁₄ N ₂ PO ₃ Cl	53.41 (53.50)	4.26 (4.19)	8.22 (8.31)	1644	1286	–	3230	907 (P–O _{ali})	1080 (O–C _{ali})
3i	188–190 (d)	45	C ₂₀ H ₁₇ N ₂ PO ₂ S	63.28 (63.14)	4.41 (4.50)	7.24 (7.36)	1620	–	797	3348	906	1221
3j	161–163 (d)	42	C ₁₉ H ₁₄ N ₂ PO ₂ SCl	56.82 (56.93)	3.58 (3.52)	6.84 (6.93)	1632	–	792	3296	959	1194
3k	48–150 (d)	39	C ₁₉ H ₁₅ N ₂ PO ₂ S	62.17 (62.28)	4.06 (4.12)	7.68 (7.64)	1642	–	768	3362	975	1218

Note: Et₃N/triethyl amine; THF, tetrahydrofuran.

^aRecrystallized from aqueous 2-propanol.

TABLE 2 ¹H and ³¹P NMR Spectral Data^a of **3** (δ from TMS)

Compound	<i>H</i> -(4,6,7) ^b	<i>OAr</i> - <i>H</i>	<i>OAr</i> - <i>CH</i> ₃	<i>O</i> = <i>CC</i> ₆ <i>H</i> ₅	<i>NH</i>	³¹ P NMR ^c
3a	7.15 (s, 1H, 4-H), 6.63 (d, 8.2, 1H,6-H), 7.47 (d, 8.2, 1H,7-H)	7.20–7.59 (m, 4H)	2.22 (s, 3H)	7.02–7.84 (m, 5H)	5.91 (brs, 2H)	–5.61, –6.30
3b	7.26 (s, 1H, 4-H), 6.74 (d, 8.5, 1H, 6-H), 7.39 (d, 8.5, 1H, 7-H)	6.98–7.42 (m, 4H)	2.25 (s, 3H)	6.82–7.52 (m, 5H)	5.85 (brs, 2H)	–5.42, –6.04
3c	6.78 (s, 1H, 4-H), 6.26 (d, 8.1, 1H, 6-H), 7.02 (d, 8.1, 1H, 7-H)	7.40–7.65 (m, 5H)	–	7.21–7.56 (m, 5H)	–	–5.53, –5.95
3d	7.03 (s, 1H, 4-H), 6.67 (d, 8.2, 1H, 6-H), 7.18 (d, 8.2, 1H, 7-H)	7.62–7.68 (m, 4H)	–	7.53–7.58 (m, 5H)	5.27	–5.30, –5.97
3e	7.09 (s, 1H, 4-H), 6.82 (d, 8.3, 1H, 6-H), 7.12 (d, 8.3, 1H, 7-H)	7.24–7.53 (m, 4H)	–	6.97–7.44 (m, 5H)	5.45 (brs, 2H)	–5.16, –6.08
3f	7.02 (s, 1H, 4-H), 6.94 (d, 8.4, 1H, 6-H), 7.26 (d, 8.4, 1H, 7-H)	7.31–7.46 (m, 3H)	2.04 (s, 3H, 2''-CH ₃) 2.16 (s, 3H, 5''-CH ₃)	7.17–7.39	6.20	15.26, 15.48
3g	6.73 (s, 1H, 4-H), 6.60 (d, 8.2, 1H, 6-H), 6.89 (d, 8.2, 1H, 7-H)	7.54–7.84 (m, 3H)	2.15 (s, 6H, 2'',6'' -(CH ₃) ₂)	(7.16–7.40) (m, 5H)	6.32 (brs, 2H)	15.72, 15.88
3h	7.15 (s, 1H, 4-H), 6.53 (d, 8.3, 1H, 6-H), 6.96 (d, 8.3, 1H, 7-H)	OCH ₂ CH ₂ Cl 3.42–4.22 (m, 4H)	–	7.21–7.51 (m, 5H)	5.62 (brs, 2H)	–
3i	7.21 (s, 1H, 4-H), 6.81 (d, 8.2, 1H, 6-H), 7.37 (d, 8.2, 1H, 7-H)	7.41–7.72 (m, 4H)	2.55 (s, 3H)	7.31–7.67 (m, 5H)	5.37 (brs, 2H)	–
3j	7.32 (s, 1H, 4-H), 7.08 (d, 8.4, 1H, 6-H), 7.68 (d, 8.4, 1H, 7-H)	7.38–7.84 (m, 4H)	–	7.18–7.56 (m, 5H)	5.85 (brs, 2H)	–
3k	6.82 (s, 1H, 4-H), 6.49 (d, 8.1, 1H, 6-H), 7.34 (d, 8.1, 1H, 7-H)	7.80–7.74 (m, 5H)	–	7.34–7.49 (m, 5H)	5.24 (brs, 2H)	–

^aRecorded in DMSO-*d*₆.^bValues in parentheses are coupling constants; *J* in Hz.^c³¹P chemical shifts were expressed in δ from 85% H₃PO₄ as external standard.

Varian Gemini 400 MHz NMR spectrometer operating at 162 MHz. All spectra were recorded using DMSO-*d*₆ with TMS as the internal reference for ¹H and ¹³C and 85% H₃PO₄ external reference for ³¹P NMR.

3,4-Diaminobenzophenone (**1**) was procured from Aldrich Chemical Company, Milwaukee, and was used without further purification.

2-(2''6''-Dimethylphenoxy)-2,3-dihydro-5-benzoyl-1*H*-1,3,2-benzodiazaphosphole 2-oxide (**3g**)

A solution of 2'',6''-dimethylphenyl phosphoro dichloridate (**2g**, 0.476g, 0.002 mol) in 25 ml of dry toluene was added dropwise over a period of 20 min to a stirred solution of 3,4-diaminobenzophenone

(**1**, 0.424 g, 0.02 mol) and triethylamine (0.404 g, 0.04 mol) in 30 ml of dry toluene and 10 ml of tetrahydrofuran. After completion of the addition, the temperature of the reaction mixture was slowly raised to room temperature and stirred for 2 h. Later the reaction mixture was heated to 50–60°C and maintained for 4 h with stirring. The completion of the reaction was monitored by TLC analysis. The condensed product, which was insoluble in toluene, was separated by filtration. The gummy solid obtained was washed with water to remove residual traces of triethylamine hydrochloride and dried. The solid obtained was recrystallized from aqueous 2-propanol: yield 57%, m.p. 293–295°C. Anal. calcd for C₂₁H₁₉N₂PO₃ (378.3634): C, 66.66; H, 5.06; N, 7.40. Found: C, 66.75; H, 5.18; N, 7.32.

TABLE 3 ^{13}C NMR Data of Compounds of **3**^a (δ from TMS)^b

Carbon Atoms	3a	3b	3c	3d	3g	3i	3j
C-1'	139.56	139.42	139.23	139.78	138.24	138.98	138.46
C-2'-6'	129.49	129.44	129.35	129.13	128.94	129.22	128.92
C-4'	131.15	131.12	131.46	131.18	131.79	131.67	131.72
C-3',5'	129.43	129.48	129.40	129.13	128.94	129.22	129.41
C-10	194.18	194.26	193.96	194.35	195.04	193.74	194.23
C-4	113.24 (12.4)	113.18 (12.5)	116.12 (12.2)	112.67 (12.0)	113.30 (12.6)	114.67 (12.8)	114.28 (12.5)
C-5	139.56	140.12	139.62	139.78	139.40	138.98	139.42
C-6	126.95	126.78	127.20	125.52	127.00	127.31	127.26
C-7	118.10 (12.4)	118.06 (12.5)	119.21 (12.2)	116.22 (12.0)	119.44 (12.6)	122.20 (12.8)	122.02 (12.5)
C-8	142.54	142.62	143.51	141.73	142.80	144.73	144.36
C-9	139.56	139.54	139.41	139.78	139.40	138.98	139.04
C-1''	150.45	150.49	150.54	149.86	153.25	151.67	152.24
C-2''	129.43	129.42	126.48	131.18	131.79	122.20	121.92
C-3''	130.70	130.68	130.79	128.40	129.35	129.18	129.24
C-4''	125.47	125.42	126.08	125.52	124.46	123.10	122.86
C-5''	126.95	126.86	127.02	123.37	129.35	129.78	129.62
C-6''	124.12	123.96	124.26	116.22	119.44	122.20	121.84
	16.53 (2''-CH ₃) 20.02 (4''-CH ₃)				(2''-CH ₃) 16.91 20.24 (4''-CH ₃)		
					(6''-CH ₃)		

^aNot recorded for **3e**, **3f**, **3h**, and **3k**.^bData in parentheses are coupling constants $J_{\text{p-c}}$ (in Hz).TABLE 4 Antifungal and Antibacterial Activities of Compounds **3** in Terms of Zone of Inhibition (mm)

Compound	Fungi								Bacteria	
	<i>Aspergillus flavus</i>		<i>Penicillium notatum</i>		<i>Helminthosporium anomolum</i>		<i>Fusarium oxysporum</i>		<i>Bacillus subtilis</i>	<i>Klebsiella pneumoniae</i>
	10 μg	20 μg	10 μg	20 μg	10 μg	20 μg	10 μg	20 μg	10 μg	10 μg
3a	–	–	–	–	10	18	5	12	–	–
3b	–	–	–	–	11	20	4	9	–	–
3c	–	–	6	12	6	15	10	14	–	–
3d	10	22	2	4	10	20	–	–	14	16
3e	8	15	5	9	9	17	–	–	11	15
3i	14	24	12	21	12	18	4	8	10	12

Empty entries indicate no activity.

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REFERENCES

- [1] Cremllyn, R. J. W.; Wakeford, D. H. Topics in Phosphorus Chemistry; John Wiley & Sons: New York, 1976; Vol. 8.
- [2] Anuradha, K.; Rajasekhar, A.; Reddy, T. K. K.; Reddy, C. D. Ind Phytopatholo 1991, 44, 414.
- [3] Arnold, H.; Bourseaux, F.; Brock, N. Arzenimittel Forsch 1961, 11, 143.
- [4] Zhadanov, R. I.; Buina, W. A.; Kapitanova, N. G. Nuretdinov, I. A. Synthesis 1979, 269.
- [5] Zimmer, H.; Sill, A. Prog Drug Res 1964, 5, 150.
- [6] Friedman, O. M.; Papanastassiou, Z. B.; Levi, R. S. J Med Chem 1963, 6, 82.
- [7] Kuranari, M. Jpn Patent 26, 819, 1965, (Chugai Pharmaceutical Co.); Chem Abstr 1966, 64, 9737.
- [8] Nyquist, R. A. Spectrochim Acta 1963, 19, 713.
- [9] Thomas, L. C.; Chittenden, R. A. Spectrochim Acta 1964, 20, 489.
- [10] Thomas, L. C.; Chittenden, R. A. Chem Soc (London) 1961, 1913.
- [11] Thomas, L. C. The Interpretation of the Infrared Spectra of Organophosphorus Compounds; Heyden: London, 1974.

- [12] Reddy, C. D.; Reddy, R. S.; Reddy, M. S.; Krishnaiah, M.; Berlin, K. D.; Sunthankar P. Phosphorus Sulfur Silicon 1991, 62, 1.
- [13] Rao, L. N.; Reddy, C. D.; Berlin, K. D. Heterocyclic Chem 2000, 37, 275.
- [14] Venugopal, M.; Reddy, C. D.; Nagaraju, C.; Berlin, K. D. Heterocyclic Commun 2000, 63, 253.
- [15] Muller, N.; Lauterbur, P. C.; Goledeenron J. J Am Chem Soc 1956, 78, 3557.
- [16] Vanwazer, J. R.; Callis, C. F.; Shoolery, J. N.; Jones, R. C. J Am Chem Soc 1956, 78, 5715.
- [17] Remirez, F.; Prasad, V. A. U.; Maracek, J. F. J Am Chem Soc 1979, 96, 7269.
- [18] Uma Maheswari Devi, P.; Reddy, P. S.; Usha Rani, N. R.; Reddemma, P. Eur J Plant Pathol 2000, 106, 857-865.
- [19] Vincent, J. C.; Vincent, H. W. Proc Soc Exp Biol Med 1944, 55, 162.